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# Phosphatase activity of Poa pratensis seeds\*

III. Effect of fluoride, citrate, urea and other substances on the activity of acid phosphatase Ia<sub>2</sub> and Ia<sub>3</sub>

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#### Abstract

Effects of fluoride, citrate, urea and other substances on the activity of acid phosphatase  $a_2$  and  $a_3$  toward p-nitrophenylphosphate and phenylphosphate were investigated. Both enzymes were inhibited by fluoride, p-chloromercuribenzoate and oxalate. Fluoride inhibited acid phosphatase  $a_2$  noncompetitively with p-nitrophenylphosphate, whereas acid phosphatase  $a_3$  showed inhibition of mixed type. Hydrolysis of phenylphosphate by both acid phosphatases was activated by citrate. Cytosine and uridine inhibited the activity of phosphatase  $a_2$  toward p-nitrophenylphosphate and phenylphosphate, but no effect was observed in case of acid phosphatase  $a_3$ . After 30 min. incubation with 4 M urea both enzymes lost about  $30^0/_0$  of activity.

### INTRODUCTION

Molecular forms of acid phosphatases occuring in fungi and higher plants differ with respect to their location in cell structures (Wakao et al. 1975, Chen et al. 1975, Parish 1972), molecular weight (Kruzel 1977), and catalytic properties (Verjee 1969).

Isolation, purification, and preliminary characteristic of the two forms of acid phosphatase of Poa pratensis seeds have been given in the previous paper (Lorenc-Kubis et al. 1975). It has been shown that acid phosphatase  $a_2$  and  $a_3$  is characterized by low substrate specificity and inhibition of the type caused by zinc ions. The present work con-

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stitutes a continuation of studies on these enzymes. It was aimed at determining the type of inhibition caused by fluoride ions, and at studying the effect of citrate, urea and other substances on the activity of both enzymes.

## MATERIAL and METHODS

Studies were carried out on acid phosphatase  $a_2$  and  $a_3$  isolated from seeds of meadow-gras (*Poa pratensis*), according to the method described in the previous paper (Lorenc-Kubis et al. 1975).

Specific activity of each enzyme amounted to 10 units/mg of protein. Proteins were determined with turbidimetric tannin micromethod according to Mejbaum-Katzenellenbogen (1955). Activity of acid phosphatase was determined measuring p-nitrophenyl or inorganic phosphate-liberated either from p-nitrophenylphosphate or from phenylphosphate after 10 min and 30 min in  $37^{\circ}\text{C}$  pH 5.1, according to the method described previously (Lorenc-Kubis et al. 1975).

One unit of phosphatase activity represents 1  $\mu$ mol of p-nitrophenyl or inorganic phosphate liberated from the substrate during 1 min in 37°C in 0.1 M acetate buffer, at pH 5.1.

#### RESULTS and DISCUSSION

Table 1 presents the effect of various substances on the activity of acid phosphatase  $a_2$  and  $a_3$  toward the substrates p-nitrophenylphosphate and sodium phenylphosphate. Fluoride ions inhibited in  $70^{\circ}/_{\circ}$  the activity of acid phosphatase  $a_2$  and  $a_3$  toward p-nitrophenylphosphate (Lorenc-Kubis et al. 1975), and in  $80^{\circ}/_{\circ}$  the activity of each enzyme toward phenylphosphate. Fluoride ions constitute uncompetitive inhibitors for many acid phosphatases of plant origin (Yoshida and Tamiya 1971, Uehara et al. 1974).

Inhibition of the activity of acid phosphatase  $a_2$  by fluoride ions was of a noncompetitive character, whereas in case of acid phosphatase  $a_3$  — of the mixed type (Fig. 1 and 2). Estimated values of  $K_m$  and  $K_i$  for acid phosphatase  $a_2$  amount to  $1.66 \times 10^{-3}$ , and for acid phosphatase  $a_3$  to:  $K_m$   $6.66 \times 10^{-4}$  and  $K_i$   $1.54 \times 10^{-3}$ . Also p-chloromercuribenzoate and oxalate ions inhibted the enzymes under study (Tab. 1) In the presence of  $2 \times 10^{-5}$  M pCMB acid phosphatase  $a_2$  retained 62-70% of its total activity, and acid phosphatase  $a_3$  — 70-80%. Decrease of the activity of both enzymes under the effect of pCMB most probably shows that free sulfhydryl groups of the enzymes participate in the

hydrolysis of the substrates. Similar result of the inhibition caused by pCMB was found in case of two acid phosphatases  $E_{\rm II}$  and  $E_{\rm III}$  isolated from wheat germs (Verjee 1969), as also in case of acid phosphatase Ib and Ic of rice bran (Igaue et al. 1975).

Cytosine, as well as uridine, decrease the activity of acid phosphatase  $a_2$  toward both substrates under study. On the other hand guanine inhibits the activity of acid phosphatase  $a_2$  toward p-nitrophenylphosphate, and of acid phosphatase  $a_3$  toward phenylphosphate, in about  $20^{6}/_{0}$ . Ribose activates the effect of acid phosphatase  $a_2$  toward p-nitrophenylphosphate (Tab. 1).

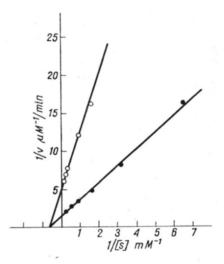


Fig. 1. Effect of fluoride ions on acid phosphatase a<sub>2</sub> activity, substrate — p-nitrophenylphosphate

The reaction mixture contained in a total volume of 2,0 ml 0,1 M acetate buffer, pH 5,1; 0,035 unit of phasphatase  $a_2$ ,  $1\times10^{-3}$  M NaF and indicated amount of p-nitrophenylphosphate. Incubation was made for 10 min. at 37°. Reaction was stopped by addition of 10 ml 0,1 N NaOH, and the absorbance of the solution was measured at 410 nm.

• — • — with no inhibitor; • — • — with inhibitor

Fig. 3 presents the effect of citrate on the activity of acid phosphatase a<sub>2</sub> and a<sub>3</sub> toward p-nitrophenylphosphate and phenylphosphate. Some differences were noted in the activity of both enzymes under study. In the presence of 10 — 40 mM citrate the activity of acid phosphatase a<sub>2</sub> and a<sub>3</sub> toward phenylphosphate increased, while the activity of these enzymes toward p-nitrophenylphosphate remained almost unchanged.

### ' Table 1

Effect of various compounds on the activity of acid phosphatase a<sub>2</sub> and a<sub>3</sub> toward p-nitrophenylphosphate and phenylphosphate

Reaction mixture contained in a total volume of 2,0 ml 0,1 M acetate buffer, pH 5,1: 0,035 unit of enzyme, 10 µmoles of p-nitrophenylphosphate or phenylphosphate and indicated amount of various compounds. Incubation was made for 10 or 30 min. 37°. The reaction was stopped by addition of 2 ml of 5% trichloroacetic acid. The liberated inorganic phosphate was determined by the method of Fiske and Subbarow.

Compound added		Relative activity %			
		AcPh a <sub>2</sub>		AcPh a <sub>3</sub>	
	М	p-NPP	Phenyl- phosphate	p-NPP	Phenyl- phosphate
NaF	1×10 <sup>-2</sup>	27	18	31	25
p-nitrophenol	1×10 <sup>-3</sup>	100	100	100	100
Na-citrate	1×10 <sup>-2</sup>	105	117	100	113
K-oxalate	1×10 <sup>-2</sup>	65	80	93	94
p-CMB	2×10 <sup>-5</sup>	62	70	71	80
urea	1×1	100	100	100	100
adenine	1×10 <sup>-3</sup>	100	93	100	92
cytosine	1×10 <sup>-3</sup>	78	91	100	100
uridine	$1 \times 10^{-3}$	86	89	100	92
guanine	1×10 <sup>-3</sup>	84	100	100	81
l-arabinose	1×10 <sup>-3</sup>	100	100	100	93
ribose	1×10 <sup>-2</sup>	107	92	120	100
none		100	100	100	100

AcPh a2, AcPh a3 — Acid phosphatase a2 and a3, p-NPP — p-nitrophenylphosphate. p-CMB — p-chloromercuribenzoate

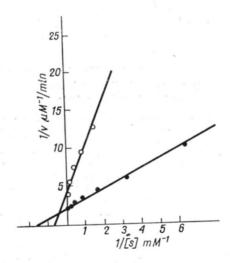


Fig. 2. Effect of fluoride ions on acid phosphatase a<sub>3</sub> activity, substrate — p-nitrophenylphosphate

The assays were carried out as described in the legend to Fig. 1

with no inhibitor; o—o — with inhibitor.

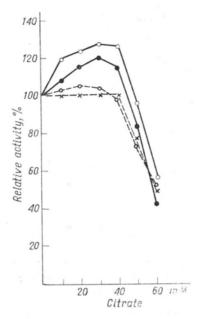


Fig. 3. Effect of Na-citrate on p-nitrophenylphosphate and phenylphosphate hydrolysis by acid phosphatase a2 and a3

Reaction mixture contained in a total volume of 2.0 ml 0.1 M acetate buffer pH 5.1:0.035 unit of enzyme, 10 µmoles of p-nitrophenyl phosphate or phenylphosphate and indicated amound of Na-citrate. Incubation was made for 30 min. 37°. The liberated inorganic phosphate was determined after Fiske and Subbarow. p-nitrophenyl phosphate: o——o acid phosphatase a2; x———x acid phosphatase a3 phenylphosphate: o——o acid phosphatase a2

The effect of urea on the activity of acid phosphatase  $a_2$  and  $a_3$  is presented in Fig. 4. It was found that urea in concentrations of 4-6 M caused inactivation of both enzymes. In the presence of 4 M urea after 30 min incubation in pH 5.1 the activity of the enzymes lowered to about  $70^{\circ}/_{\circ}$ , and in the presence of 6 M urea — to about  $30^{\circ}/_{\circ}$ . Prolonged time of enzyme incubation in the presence of 4 M urea (Fig. 5) resulted in a significant decrease of the activity of phosphatase  $a_3$ , with simultaneous slight decrease of the activity of acid phosphatase  $a_2$ . Lowered activity in the presence of urea was noted also in case of acid phosphatase of yeasts (W a t o r e k and M o r a w i e c k a 1977), as also in case of three acid phosphatases of wheat germs (V e r j e e 1969) and acid phosphatase of potatoes (K r u z e l 1977). It may be assumed that both forms of acid phosphatase ( $a_2$  and  $a_3$ ) of *Poa pratensis* seeds, similarly as acid phosphatase of potatoes, possess subunital structure or else they form a complex enzymatic system.

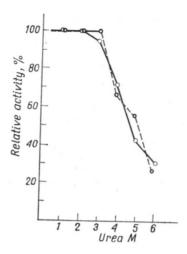


Fig. 4. Effect of urea on acid phosphatase a<sub>2</sub> and a<sub>3</sub> activity, substrate — p-nitrophenylphosphate

The acid phopshatase  $a_2$  and  $a_3$  (0,3 — 0,4 units) were incubated with 1 — 6 M urea in 0,1 M acetate buffer, pH 5,1, 37°. After 30 min. aliquots were withdrawn and activity toward p-nitrophenylphosphate was assayed under standard conditions

o----o phosphatase a2; o---o phosphatase a2

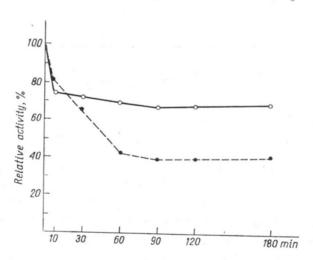


Fig. 5. Effect of 4 M urea on acid phosphatase  $a_2$  and  $a_3$  activity, substrate — p-nitrophenylphosphate

The acid phosphate  $a_2$  and  $a_3$  (0,3 - 0,4 unit) were incubated with 4 M urea in 0,1 M acetate buffer, pH 5,1 at 37°. After various times aliquots were withdrawn and the activity toward p-nitrophenyl phosphate was assayed under the standard conditions

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# Aktywność fosfatazowa nasion wiechliny łąkowej (Poa pratensis)

III. Wpływ fluorku, cytrynianu, mocznika oraz innych substancji na aktywność fosfatazy kwaśnej Ia<sub>2</sub> i Ia<sub>3</sub>

# Streszczenie

Zbadano wpływ fluorków, cytrynianu, mocznika i innych substancji na aktywność fosfatazy kwaśnej  $a_2$  i  $a_3$  nasion wiechliny łąkowej (*Poa pratensis*). Badania przeprowadzono wobec dwóch substratów: p-nitrofenylofosforanu i fenylofosforanu sodu. Stwierdzono, że inhibitorami obu enzymów są fluorki, szczawiznoraz p-chlorortęciobenzoesan. Jony fluorkowe okazały się inhibitorami niekompetycyjnymi dla fosfatazy kwaśnej  $a_2$  podczas gdy hamowanie aktywności fosfatazy kwaśnej  $a_3$  ma charakter inhibicji mieszanej. Cytrynian aktywował oba enzymy wobec fenylofosforanu, nie wywierał natomiast wpływu na hydrolizę p-nitrofenylofosforanu. Cytozyna i urydyna obniżają aktywność fosfatazy kwaśnej  $a_2$  wobec obu substratów, bez wpływu pozostają natomiast na aktywność fosfatazy kwaśnej  $a_3$ . Mocznik w stężeniu 4 M inaktywuje oba enzymy w około  $30^0/_0$  po minutowej inkubacji.